

*Answer these questions on page 13 of your Interactive Notebook*

**Naming Ionic Compounds Quick Practice:**

1. NaCl: \_\_\_\_\_ + \_\_\_\_\_ + ide: \_\_\_\_\_
2. LiO: \_\_\_\_\_ + \_\_\_\_\_ + ide: \_\_\_\_\_
3. MgF: \_\_\_\_\_ + \_\_\_\_\_ + ide: \_\_\_\_\_
4. KCl: \_\_\_\_\_ + \_\_\_\_\_ + ide: \_\_\_\_\_

5. Lithium Fluoride: \_\_\_\_\_
6. Magnesium Chloride: \_\_\_\_\_
7. Potassium Oxide: \_\_\_\_\_

***Extension activity:***

**Create your own ionic compounds! Choose a metal and non-metal pair. Then write down the formula and the compound name.**